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KWAZULU-NATAL PROVINCE

EDUCATION REPUBLIC OF SOUTH AFRICA

PHYSICAL SCIENCES SCOPE FOR EXAMINATIONS 2024: GRADE 12

MARCH CONTROLLED TEST (100 marks)				
PAPER	TOPICS	MARK	DURATION	
	Mechanics Momentum & Impulse: ✓ Momentum, Newton's second law expressed in terms of Momentum, ✓ Conservation of momentum and Elastic and Inelastic collisions. Vertical Projectile Motion in One Dimension (1D): ✓ represented in words, diagrams, equations and graphs Grade 11 Work ✓ Newton's Laws (Newton 1, 2, 3) ✓ Newton's Law of Universal Gravitation) ✓ Applications of Newton's Laws	50		
ONE PAPER ONLY	 Matter and Materials Organic molecules: ✓ Organic molecular structures - functional groups, saturated and unsaturated structures, isomers ✓ IUPAC naming and formulae ✓ Structure physical property relationships ✓ Applications of organic chemistry - Substitution, addition and elimination. (ONLY alkanes, alkenes, alkynes, alcohols, haloalkanes, carboxylic acids, and esters) Grade 11 Work ✓ Inter-molecular forces NB: Quantitative aspects of chemical change may be assessed across chemistry. 	50	2 hours	

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	JUNE EXAMINATION (2 PAPERS)		
PAPER	TOPICS	MARKS	DURATION
PAPER 1	Momentum & Impulse: ✓ Momentum, Newton's second law expressed in terms of Momentum, ✓ Conservation of momentum and Elastic and Inelastic collisions. Vertical projectile motion in one dimension (1D): ✓ represented in words, diagrams, equations and graphs Grade 11 Work ✓ Newton's Laws (Newton 1, 2, 3 and Universal Gravitational Law) ✓ Applications of Newton's Laws Work, Energy & Power: ✓ Definition of Work, ✓ Work Energy Theorem ✓ Conservation of energy with non-conservative forces ✓ Power Waves, Sound and Light: Doppler Effect ✓ Doppler Effect Calculations ✓ Application with sound and ultrasound, ✓ Application with light – blue shifts and red shifts in the universe	100	2 Hours
PAPER 2	(Evidence for the expanding universe). Organic molecules ✓ Organic molecular structures - functional groups, saturated and unsaturated structures, isomers ✓ IUPAC naming and formulae ✓ Structure physical property relationships ✓ Applications of organic chemistry - Substitution, addition and elimination. (ONLY alkanes, alkenes, alkynes, alcohols, haloalkanes, carboxylic acids, and esters) ✓ Calculations (Quantitative aspects of chemical changes) Chemical Change Rate and Extent of Reaction: ✓ Rates of reaction ✓ factors affecting rate (nature of reacting substances, concentration [pressure for gases], temperature and presence of a catalyst) ✓ Measuring rates of reaction; Mechanism of reaction and of catalysis ✓ Calculations (Quantitative aspects of chemical changes) Chemical Equilibrium: ✓ Chemical equilibrium ✓ Factors affecting equilibrium ✓ Equilibrium constant ✓ Application of equilibrium principles. Acids and Bases: ✓ Definitions, strong and weak, concentrated and dilute, conjugate acid-base pairs, neutralization, titrations ✓ PH calculations ✓ Dilution and neutralization ✓ Acid-base reactions ✓ Calculations (Quantitative aspects of chemical changes)	100	2 hours

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PREPARATORY EXAMINATIONS

FORMAT OF QUESTION PAPERS

Paper 1: Physics 3 hours	Paper 2: Chemistry 3 hours
SECTION A:	SECTION A:
Multiple-choice questions	Multiple-choice questions
SECTION B: Conceptual questions assessing all themes	SECTION B: Conceptual questions assessing all themes
Total: 150 marks	Total: 150 marks

Note: Full Papers will be written, including selected examinable Grades 10 & 11 Topics (p.149 CAPS Document)

MARK ALLOCATION PER KNOWLEDGE AREA

	Knowledge Area	Theme	Marks	
		Newton's Laws (1,2,3) and Law of Gravitation	65	
	Mechanics	Momentum (1D), Impulse and change in momentum		
	(±43,3%)	Vertical projectile motion (1D)		
		Work, power and energy		
7	Waves, sound and light	Doppler effect		
PAPER	(±10%)			
		Electrostatics (Grade 11)		
	Electricity and Magnetism	Electric circuits (Grades 11 & 12)	55	
	(±36,7%)	Electrodynamics Grade 12		
		Electromagnetic radiation Grade 12		
	Matter and materials (10%)	Optical phenomena and properties of materials	15	
		TOTAL	150	
PAPER 2	Knowledge Area	Theme	Marks	
	Matter and materials (±38,7%)	Organic molecules	58	
	-	Rate and extent of reaction		
	Chemical change	Chemical Equilibrium		
	(±61,3%)	%) Acids and Bases		
		Electrochemistry and Electrochemical cells		
		TOTAL	150	

COGNITIVE LEVELS

Cognitive level description	Weighting %		
Cognitive level description	Paper 1	Paper 2	
Remembering	15	15	
Understanding	35	40	
Applying and Analysing	40	35	
Evaluating	10	10	