

NATURAL SCIENCES

GRADE 8

INFORMAL TEST 2.2

TOPIC: Particle model of matter, Change of state

Density, mass, volume, Density and states of matter

MARKS: 15

Name and Surname: _____ Gr. 8 _____ Date: _____

SECTION A

QUESTION 1

1.1 Match the description in COLUMN A with the correct term in COLUMN B. Write only the LETTER (A - E) next to the question number (1.1.1 – 1.1.3) in the open spaces provided below.

COLUMN A		COLUMN B	
1.1.1	The state of matter that completely fills the container in which it is placed.	A	Diffusion
1.1.2	The process of converting a liquid into a solid.	B	Solid
1.1.3	The movement of particles from high to low concentration.	C	Gas
		D	Freezing

ANSWERS: 1.1.1 _____ 1.1.2 _____ 1.1.3 _____ (3)

1.2 Give the correct word OR term for each of the following statements.

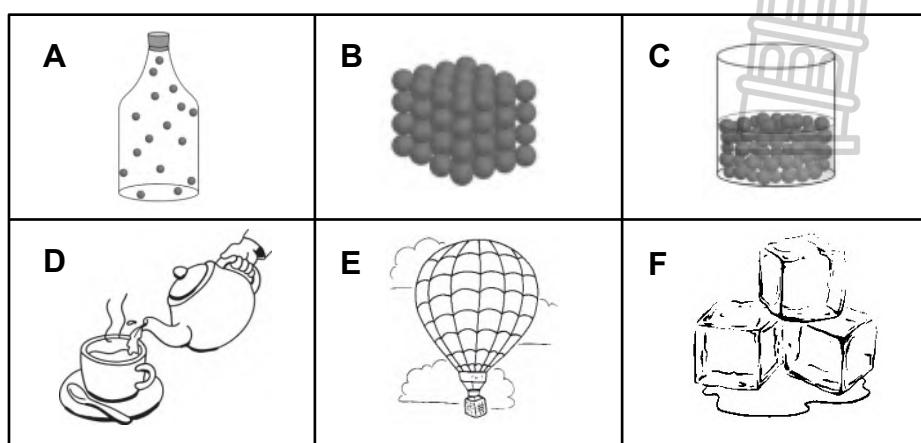
1.2.1 The mass per unit volume of a substance. Answer: _____ (1)

1.2.2 A measure of the amount of matter in an object. Answer: _____ (1)
[5]

SECTION B

QUESTION 2

2.1 Study diagrams A to F below. Diagrams A, B and C represent different states of matter. In diagram D, tea is poured into a cup. Diagram E shows a hot air balloon and diagram F shows three cubes of ice. Answer the questions that follow.



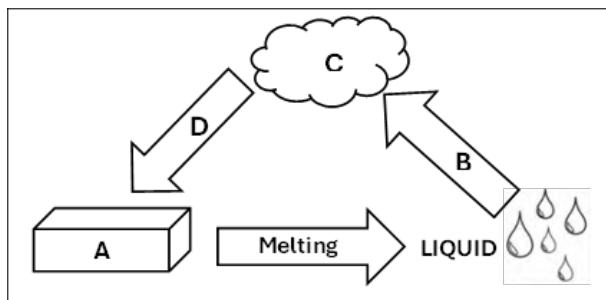
2.1.1 Give the LETTER of the TWO diagrams showing a LIQUID: _____ and _____ (2)

2.1.2 Give a reason VISIBLE in the diagram why diagram B represents the solid state of matter.

_____ (1)

2.1.3 Select the state of matter (A, B, or C) that most accurately shows the arrangement of particles within the hot air balloon (E). _____ (1)

2.2 The diagram below shows the three states of matter and the processes taking place when a substance changes from one state (phase) to another.



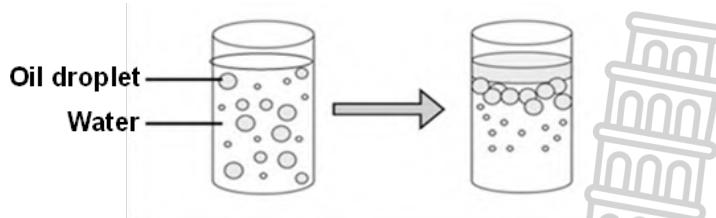
2.2.1 Name the state of matter represented by LETTER A. _____ (1)

2.2.2 Which arrow (B or D) indicates the process of evaporation? _____ (1)

2.2.3 Explain what happens during melting.

_____ (1)

2.3 Oil with a density of $0,85 \text{ g/cm}^3$ is mixed well with water, which has a density of 1 g/cm^3 . The mixture is left to stand for a few minutes and the observation made, is shown below.



2.3.1 Describe what is observed after the mixture was left to stand for a few minutes.

_____ (2)

2.3.2 Give a reason for your answer in 2.3.1.

_____ (1)
[10]

TOTAL MARKS: 15

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MEMORANDUM

SECTION A

QUESTION 1

1.1.1 C ✓	(1)
1.1.2 D ✓	(1)
1.1.3 A ✓	(1)
1.2.1 Density ✓	(1)
1.2.2 Mass ✓	(1)
	[5]

SECTION B

QUESTION 2

2.1.1 C✓ and D✓ (in any order)	(2)
2.1.2 NOTE: Only accept what is VISIBLE in diagram B. Particles are closely packed. ✓ Particles are in a regular arrangement. ✓ Particles have (very) small spaces ✓ between them. (ANY ONE)	(1)
2.1.3 A✓	(1)
2.2.1 Solid✓	(1)
2.2.2 B✓	(1)
2.2.3 A solid changes into a liquid. ✓	(1)
2.3.1 The oil droplets rise (go up)✓ and float on the water.✓ OR The oil and water separate, with the oil on top ✓ and the water at the bottom. ✓	(2)
2.3.2 Oil is less dense ✓ than water.	(1)

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